



Physiology

Dissipation of excess photosynthetic energy contributes to salinity tolerance: A comparative study of salt-tolerant *Ricinus communis* and salt-sensitive *Jatropha curcas*[☆]



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ABSTRACT

The relationships between salt tolerance and photosynthetic mechanisms of excess energy dissipation were assessed using two species that exhibit contrasting responses to salinity, *Ricinus communis* (tolerant) and *Jatropha curcas* (sensitive). The salt tolerance of *R. communis* was indicated by unchanged electrolyte leakage (cellular integrity) and dry weight in leaves, whereas these parameters were greatly affected in *J. curcas*. The leaf Na⁺ content was similar in both species. Photosynthesis was intensely decreased in both species, but the reduction was more pronounced in *J. curcas*. In this species biochemical limitations in photosynthesis were more prominent, as indicated by increased C_i values and decreased Rubisco activity. Salinity decreased both the V_{cmax} (*in vivo* Rubisco activity) and J_{max} (maximum electron transport rate) more significantly in *J. curcas*. The higher tolerance in *R. communis* was positively associated with higher photorespiratory activity, nitrate assimilation and higher cyclic electron flow. The high activity of these alternative electron sinks in *R. communis* was closely associated with a more efficient photoprotection mechanism. In conclusion, salt tolerance in *R. communis*, compared with *J. curcas*, is related to higher electron partitioning from the photosynthetic electron transport chain to alternative sinks.

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Introduction

Salinity affects CO₂ photosynthetic assimilation initially causing decreased stomata conductance, which is induced primarily by the osmotic effects of salt excess in roots, involving ABA transport and resulting in low CO₂ supply to Rubisco. In a second phase, salinity might cause biochemical effects on photosynthesis that can restrict both the photochemical and Calvin cycle reactions (Duarte et al., 2013). Impairment in CO₂ assimilation induces an imbalance in the overall photosynthetic process, particularly in the presence of high light intensity. Under these conditions, excess electron from the photochemical phase is accumulated in thylakoid membranes, which causes photoinhibition (Parvaiz and Satyawati, 2008). Plants display several strategies to protect the photosystems against photoinhibition and photodamage (Takahashi and Badger, 2011). These protective mechanisms are associated with several reactions at photochemical level (down-regulation in light harvesting, excess

energy dissipation by non-photochemical quenching, increase in water-water cycle and cyclic electron flow) and other important biochemical pathways such as the ascorbate-glutathione cycle, photorespiration and, nitrate and ammonia assimilation (Osmond et al., 1997; Niyogi, 1999; Flexas and Medrano, 2002; Takahashi and Murata, 2008). The role of the efficiency of these processes in salt tolerance is possibly species-dependent but the involved mechanisms are actually not understood.

Nitrate assimilation is very important to plant growth, and it is a process very sensitive to salinity (Masclaux-Daubresse et al., 2010). The overall nitrate assimilation pathway is closely linked to photosynthesis because nitrite and ammonia assimilation occur inside chloroplasts and they are strong consumers of electrons from reduced ferredoxin (Foyer and Noctor, 2002). Nitrate and CO₂ assimilation are two processes that are coordinately regulated by mechanisms that remain not completely elucidated (Foyer and Noctor, 2002). There is an important open question: which process is most sensitive to salinity: nitrate or CO₂ assimilation? Nitrate reduction is catalyzed by nitrate reductase and occurs in the cytosol, producing nitrite that enters into the chloroplasts (Silveira et al., 2001). Nitrate reduction consumes two electrons from NAD(P)H, which might be supplied indirectly by the photosynthetic electron transport.

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Nitrite reduction is catalyzed by nitrite reductase into chloroplasts, consuming six electrons from reduced ferredoxin (Sivasankar and Oaks, 1996). This reaction is a strong alternative sink for the photosynthetic electron transport (PET) chain, recovering oxidized ferredoxin to accept new electrons from PSII and PSI. The ammonia produced is quickly assimilated by the GS/GOGAT cycle, which consumes two electrons from reduced ferredoxin and one ATP molecule per assimilated ammonia (Foyer and Noctor, 2002). Some studies have revealed that this process might mitigate the negative effects caused by salinity in *Jatropha curcas* (Aragão et al., 2012). Moreover, the amino acid metabolism in chloroplasts is extremely active, directly or indirectly consuming photosynthetic electrons as reducing power and ATP (Masclaux-Daubresse et al., 2010). Because of these characteristics, nitrate assimilation is frequently considered a part of the whole photosynthesis (“photosynthetic nitrate assimilation”).

Photorespiration (P_R) is another important photoprotection mechanism. It originates from the oxygenation of RuBP by Rubisco and leads to the formation of one molecule of 3-phosphoglycerate and one molecule of phosphoglycolate. The photorespiratory cycle allows the conversion of this compound into 3-phosphoglycolate by a series of reactions that involve three different compartments, chloroplasts, peroxisomes and mitochondria, which releases CO_2 and NH_3 (Bagard et al., 2008). The NH_3 released by P_R is re-assimilated via GS/GOGAT in the chloroplast consuming reduced ferredoxin and ATP. P_R has the potential to sustain photons in a non-assimilatory pathway, protecting the photosynthetic apparatus against photo-oxidation (Niyogi, 1999) and photoinhibition (Osmond et al., 1997). In addition, photorespiration is very important in biochemical recycling, particularly to nitrogenous compounds under restrictive metabolic conditions such as salinity (Foyer et al., 2009). Additionally, the nitrate assimilation could interact with P_R through the impairment in the nitrate reductase activity by restriction of NADH (Foyer et al., 2009).

J. curcas and *Ricinus communis* are species that are well adapted to arid and semiarid regions as potential crops for biofuel production (Achten et al., 2008; Reddy and Matcha, 2010). In these regions, saline soils extensively exist, causing salt stress in plants (Janmohammadi et al., 2012; Rodrigues et al., 2013). Our group has recently demonstrated that *J. curcas* is a salt-sensitive species (Silva et al., 2013). Conversely, some reports have demonstrated that *R. communis* displays some characteristics of salt tolerance such as Na^+ exclusion from leaves after long-term exposure to high NaCl concentrations (Sun et al., 2013). Therefore, because these species display contrasting acclimation to salinity, it is important to elucidate the involvement of photosynthesis in salt tolerance. Photorespiration, nitrate assimilation and cyclic electron flow are important alternative sinks for excess photosynthetic electrons. These processes isolate or in combination, could alleviate the adverse effects on photosynthesis caused by salinity (Aragão et al., 2012).

We postulated that alternative sinks to excess electron from the photochemical phase could mitigate the negative salinity effects, contributing to salt tolerance. In the current study, we tested the hypothesis that plants with a higher intensity of nitrate assimilation, photorespiration and electron cyclic flow are more tolerant to salinity. The physiological significance of these processes for salt stress is discussed.

Material and methods

Plant material and growth conditions

Jatropha curcas (L.), cultivar FT2, provided by Fazenda Tamanuá, Brazil and *Ricinus communis* (L.) seeds, cultivar BRS 149 nordestina, provided by EMBRAPA, Brazil, were employed in

the current study. The seeds were previously selected by size and weight and were germinated in sand. Fifteen days after germination, seedlings were transferred to plastic pots (4L) containing half-strength Hoagland and Arnon (1950) nutrient solution at pH 6.0. The nutrient solution was replaced every two weeks. The pH of the solution was measured every two days and adjusted when necessary. Plants were grown under natural conditions in a greenhouse situated in a semiarid region in Fortaleza city, Brazil ($3^\circ 44' 38''$ S and $38^\circ 34' 11''$ W, 31 m altitude). The average values of environmental parameters inside the greenhouse over the experimental period were as follows: maximum/minimum temperature, $39^\circ\text{C}/24^\circ\text{C}$; relative humidity, 62%; maximum photosynthetic photon flux density (PPFD), $1200\ \mu\text{mol m}^{-2}\text{s}^{-1}$; photoperiod, 12 h. The salt stress was imposed in 45-day-old plants. In this period, both species presented a similar phenological stage, characterized by three pairs of fully expanded leaves. The nutrient solution was supplied with 100 mM NaCl, and the plants were subjected to this stressful condition for 6 days, a period when stressed plants showed significant decreases in gas exchange parameters and visual symptoms of salt injury. NaCl was gradually added (50 mM d^{-1}) to the solution to avoid osmotic shock. Plants cultivated in nutrient solution without NaCl were used as controls.

Gas exchange, chlorophyll *a* fluorescence, photorespiration and *P700* absorption measurements

For evaluation of gas exchange and photochemical responses, plants were transferred to a growth chamber under the controlled conditions of temperature (29°C), VPD (1.2 kPa) and PPFD ($700\ \mu\text{mol m}^{-2}\text{s}^{-1}$). After one hour of the acclimation period in the growth chamber, the third fully expanded leaf was monitored. The net CO_2 assimilation rate (P_N), transpiration (E), stomatal conductance to water vapor (g_s) and intercellular CO_2 partial pressure (C_i) were measured using a portable infra-red gas analyzer system, equipped with an LED source and a leaf chamber (IRGA LI-6400XT, LI-COR, Lincoln, NE, USA). The P_N was measured in response to changes in the photosynthetic photon flux density (PPFD) and intercellular CO_2 partial pressure (C_i). Each one of these conditions was separately controlled in the IRGA leaf chamber. For instantaneous measurements, the PPFD was fixed at $1500\ \mu\text{mol m}^{-2}\text{s}^{-1}$, 28°C , VDP between 1.0 and 1.5 kPa, and external CO_2 was fixed at atmospheric partial pressure (38 Pa). The P_N -PPFD and P_N - C_i fitting curves were determined according to the models proposed Lieth and Reynolds (1987) and Sharkey et al. (2007), respectively. The following parameters associated with photosynthetic efficiency were determined: the maximum photosynthetic rate ($P_{N\text{max}}$), maximum Rubisco carboxylation rate (V_{cmax}), maximum rate of photosynthetic electron transport (J_{max}) and respiration in the dark (R_N).

In vivo chlorophyll *a* fluorescence was measured using a LI-6400-40 Leaf Chamber Fluorometer (LI-COR, Lincoln, NE, USA) coupled with the IRGA. The actinic light utilized for measuring the gas exchange and chlorophyll *a* fluorescence was $1500\ \mu\text{mol m}^{-2}\text{s}^{-1}$ PPFD. The fluorescence parameters were measured using the saturation pulse method (Schreiber et al., 1994) in leaves exposed to light and 30-min dark-adapted conditions. The intensity and duration of the saturation light pulse were $8000\ \mu\text{mol m}^{-2}\text{s}^{-1}$ and 0.7 s, respectively. The amount of blue light was set to be 10% of the PPFD to maximize stomatal aperture (Flexas et al., 2007). The following parameters were assessed: the maximum quantum yield of PSII [$F_v/F_m = (F_m - F_o)/F_m$] was measured under 30-min dark-adapted conditions, the effective quantum yield of PSII [$\Delta F/F_m' = (F_m - F_s)/F_m'$] was measured in leaves exposed to actinic light of $1500\ \mu\text{mol m}^{-2}\text{s}^{-1}$ PPFD. The photochemical quenching coefficient was calculated as $qP = (F_m' - F_s)/(F_m' - F_o')$,

the non-photochemical quenching coefficient was calculated as $NPQ = (F_m - F_m')/F_m'$ and the actual flux of photons from the PSII was calculated as $ETR = (\Delta F/F_m' \times PPFD \times 0.5 \times 0.84)$. To evaluate the ETR, 0.5 was used as the fraction of excitation energy distributed to PSII, and 0.84 was used as the fraction of incoming light absorbed by the leaves. The F_m and F_o parameters correspond to maximum and minimum fluorescence of dark-adapted leaves, respectively; F_m' and F_s are the maximum and steady-state fluorescence in the light-adapted leaves, respectively, and F_o' is the minimum fluorescence after the far-red illumination of the previously light-exposed leaves (Maxwell and Johnson, 2000).

The photorespiration rate (P_R) was estimated as described by Bagard et al. (2008) from the measurements of gas exchange (P_N and R_d) and chlorophyll *a* fluorescence-derived ETR parameters using the following equation: $P_R = 1/12[ETR - 4(A + R_d)]$. The redox state of the PSI primary donor, P700, was measured using a Walz DUAL-PAM 100 (Walz, Effeltrich, Germany). The following parameters were assessed: the photochemical quantum yield of PSI [$Y(I)$], the estimated electron transport rate of PSI (ETRI), the non-photochemical quantum yield of PSI [$Y(ND)$], donor-side limitation of PSI and [$Y(NA)$] a the non-photochemical acceptor site limitation of PSI (Yamori et al., 2011). The relative cyclic electron flow (CEF) was estimated by the ETRI/ETRII ratios according to Yamori et al. (2011).

Water relations, electrolyte leakage and Na^+ and K^+ determinations

The leaf relative water content (RWC) was calculated as follows: $RWC = [(FW - DW)/(TW - DW)] \times 100$, where FW is the fresh weight, TW is the turgid weight measured after 6 h of saturation in deionized water at 4 °C in the dark and DW is the dry weight determined after 48 h in an oven at 75 °C (Silveira et al., 2009). The leaf pre-dawn water potential (Ψ_w) was evaluated immediately after sampling (3:30 am) using the pressure chamber method (Scholander et al., 1965). Electrolyte leakage (membrane damage) or cellular viability was measured as described previously by Silva et al. (2010). Twenty leaf discs (1.0 cm in diameter) were placed in test tubes containing 20 mL of deionized water. Flasks were incubated in a shaking water bath at 25 °C for 12 h, and the electric conductivity in the medium (L1) was measured. Next, the discs were boiled at 95 °C for 60 min and cooled to 25 °C, and the electric conductivity (L2) was measured again. The relative electrolyte leakage (EL) was estimated using the following equation: $EL = L1/L2 \times 100$. The content of Na^+ and K^+ ions was determined by flame photometry (Ferreira-Silva et al., 2008).

Determination of nitrate, ammonia, proline and glycine betaine

The leaf nitrate and ammonium were extracted with hot water (100 °C) and were determined using the methods of Cataldo et al. (1975) and Weatherburn (1967), respectively. The proline concentration in leaves was determined according to the method of Bates et al. (1973) by the formation of the complex proline-ninhydrin measured at 520 nm via a spectrophotometer. The glycine betaine (GB) concentrations in the leaves were determined according to Grieve and Grattan (1983) by the formation of periodide crystals dissolved in 1,2-dichloroethane. The absorbance was measured at 365 nm.

Nitrate reductase and glutamine-synthetase activities

The nitrate reductase (NR, EC 1.6.6.1) activity was measured using an *in vivo* method according to Hageman and Hucklesby

(1971), described in detail by Silveira et al. (2001). The reaction mixture contained a buffer (100 mM potassium phosphate, pH 7.5), 1 mM NADH, and 250 mM KNO_3 . The assay was initiated by adding leaf extract and was conducted at 30 °C for 15 min. The reaction was stopped by the addition of 2 mL of (1:1) 1% sulfanilamide in 2.4 M HCl:0.02% (m/v) N-1-naphthyl-ethylenediamine. The A_{540} of the supernatant was measured. NR activity was expressed as $\mu\text{mol NO}_2 \text{ g}^{-1} \text{ MF h}^{-1}$.

For glutamine synthetase (GS, EC 6.3.1.2) extraction, the medium contained 100 mM potassium phosphate buffer (pH 7.4) and 1 mM EDTA. GS activity was determined using the hydroxamate biosynthetic method as described by Berteli et al. (1995). The reaction mixture consisted of 100 mM Tris-HCl buffer (pH 7.0), 3.5 mM ATP (pH 7.0), 60 mM $MgSO_4$, 20 mM hydroxylamine hydrochloride neutralized with HCl and 50 mM Na-glutamate. The reaction was started by the addition of enzyme extract, and the mixture was incubated at 30 °C for 30 min. A control was performed omitting hydroxylamine from the reaction mixture. A standard curve was constructed with γ -glutamyl hydroxamate. GS activity was expressed as $\mu\text{mol } \gamma\text{-glutamyl hydroxamate (GGH)} \text{ g}^{-1} \text{ MF h}^{-1}$.

Preparation of crude extracts and determination of enzyme activities

To prepare the crude extracts, fresh leaf samples were ground to a fine powder in liquid N_2 with a mortar and pestle and extracted with ice-cold 100 mM Tris-HCl buffer (pH 8), containing 0.1 mM EDTA, 1 mM ascorbic acid, 20% glycerol, 3% PEG 6000 and 30 mM DTT. The enzymatic extract was stored at -20 °C until the determinations of the activities, which were performed immediately. CAT (EC 1.11.1.6) activity in the crude extract was measured following the oxidation of H_2O_2 at 240 nm over 300 s in the presence of 50 mM potassium phosphate buffer (pH 7.0) containing 20 mM H_2O_2 at 30 °C (Havir and McHalen, 1987). The CAT activity was determined according to the molar extinction coefficient of H_2O_2 ($36 \text{ M}^{-1} \text{ cm}^{-1}$) and was expressed as $\mu\text{mol } H_2O_2 \text{ mg}^{-1} \text{ protein min}^{-1}$. GO (EC 1.1.3.15) activity was assayed by the formation of a glyoxylate-phenylhydrazine complex at 324 nm (Baker and Tolbert, 1966). The GO assay mixture contained 100 mM of phosphate buffer (pH 8.3), 40 mM glycolic acid, 100 mM L-cysteine and 100 mM phenylhydrazine. The reaction was started with the addition of the 1 mM FMN, and the absorbance was monitored over 300 s. GO activity was calculated using the molar extinction coefficient of the glyoxylate-phenylhydrazine complex ($17 \text{ mM}^{-1} \text{ cm}^{-1}$) and was expressed as $\mu\text{mol glyoxylate mg}^{-1} \text{ protein min}^{-1}$. Rubisco activity was measured spectrophotometrically from the rate of NADH oxidation at 340 nm (Reid et al., 1997). For the determination of the initial Rubisco activity, the extract was added to 900 μL of the assay mixture, and the reaction was initiated with the addition of 0.5 mM RuBP. For measurement of total activity, the reaction was started after 15 min of incubation of the mixture reaction in the absence of RuDP. Thereafter, 0.5 mM RuBP was added, and the total activity was measured from NADH oxidation at 340 nm. Both the initial and total Rubisco activities were expressed as $\mu\text{mol CO}_2 \text{ m}^{-2} \text{ s}^{-1}$. The Rubisco activation state was calculated using the ratio initial activity/total activity and was expressed as a percentage (%).

Statistical analysis and experimental design

The experiments were arranged in a completely randomized design in a 2×2 factorial (two plant species $\times$ two NaCl levels) and four replicates, each consisting of an individual plant in a pot. Data

Table 1

Leaf dry matter (DM), relative water content (RWC), water potential (Ψ_w) and membrane damage in leaves of *Ricinus communis* and *Jatropha curcas* plants subjected to 6 days of 100 mM NaCl (salt) or control treatment. Upper-case letters represent different means between treatments for the same species, and lower-case letters represent different means between species for the same treatment (Tukey's test; $P < 0.05$).

	<i>R. communis</i>		<i>J. curcas</i>	
	Control	Salt	Control	Salt
Leaf DM (g plant^{-1})	3.64Aa	3.15Aa	2.63Ab	1.62Bb
RWC (%)	79Aa	66Ba	70Ab	65Ba
Ψ_w (MPa)	−0.42Ba	−1.24Ab	−0.22Bb	−1.49Aa
Membrane damage (%)	31Aa	34Ab	29Ba	53Aa

were analyzed using ANOVA, and the means were compared using Tukey's test at a confidence level of 0.05.

Results

Growth, water relations and organic and solutes in *J. curcas* and *R. communis*

In the present study, *R. communis* and *J. curcas* young plants were exposed to salt stress (100 mM NaCl) for six consecutive days. Salinity promoted significant reduction in leaf dry matter in *J. curcas* but no change occurred in *R. communis*, compared with the respective controls (Table 1). The leaf relative water content and water potential decreased in both species under salt stress. In addition, salt stress increased the membrane damage (indicated by electrolyte leakage) in *J. curcas* plants (by 83%), whereas this parameter did not change in *R. communis* (Table 1).

Concerning to the inorganic and organic solute contents, salt treatment similarly increased the Na^+ content in *R. communis* and *J. curcas* (Table 2). As expected, the leaf K^+ content decreased in both species under salinity but the former species exhibited lower K^+ concentration. The proline content increased in *R. communis* and

Table 2

Na^+ , K^+ , proline and glycine betaine contents in leaves of *Ricinus communis* and *Jatropha curcas* plants subjected to 6 days of 100 mM NaCl (salt) or control treatment. Upper-case letters represent different means between treatments for the same species, and lower-case letters represent different means between species for the same treatment (Tukey's test; $P < 0.05$).

	<i>R. communis</i>		<i>J. curcas</i>	
	Control	Salt	Control	Salt
Na^+ (mmol kg^{-1} DM)	76Bb	382Aa	90Ba	410Aa
K^+ (mmol kg^{-1} DM)	250Ab	143Bb	420Aa	284Ba
Proline ($\mu\text{mol g}^{-1}$ DM)	0.43Bb	0.64Aa	0.24Bb	0.43Aa
Glycine betaine ($\mu\text{mol g}^{-1}$ DM)	50.56Bb	180.47Ab	200.67Aa	250.76Aa

J. curcas exposed to salinity by 50% and 100%, respectively, but the absolute contents in both species were very low. The glycine betaine content increased by 250% and 25% in *R. communis* and *J. curcas*, respectively, but the latter species presented a content four-fold higher compared with *R. communis* (Table 2).

Photochemical, gas exchange parameters and estimated photorespiration in *J. curcas* and *R. communis*

All leaf gas exchange parameters evaluated in this study were affected by the imposed salt stress in both species (Fig. 1). The net photosynthesis (P_N) decreased by 82% and 88% in *R. communis* and *J. curcas*, respectively, and transpiration (E) and stomatal conductance (g_s) decreased by a similar magnitude. Interestingly, the intercellular CO_2 partial pressure (C_i) decreased in *R. communis* plants exposed to salinity by 25% and, in opposition, increased by 52% in *J. curcas* (Fig. 1D).

Regarding chlorophyll *a* fluorescence parameters, the effective quantum yield of PSII ($\Delta F/F_m'$) and photochemical quenching (qP) were similar and strongly reduced in both species exposed to salinity (Fig. 2). The potential quantum yield of PSII (F_v/F_m) decreased slightly in *R. communis* plants (Fig. 2B). Non-photochemical

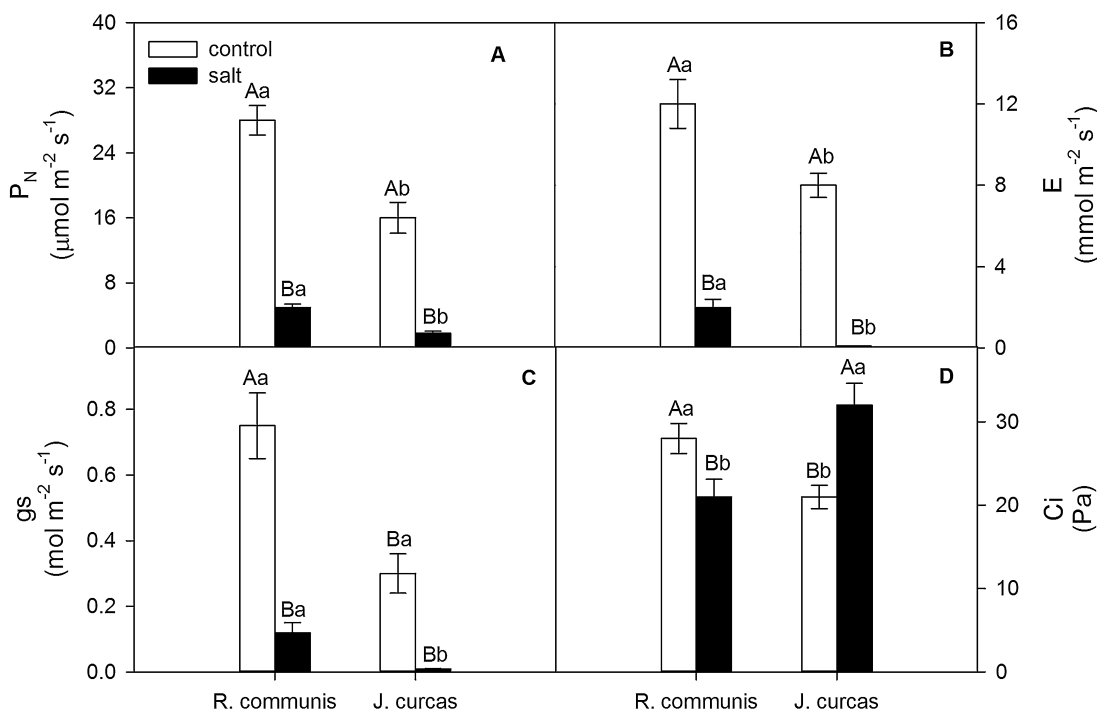


Fig. 1. Net photosynthesis (A), transpiration (B), stomatal conductance (C) and CO_2 intercellular partial pressure (D) in *R. communis* and *J. curcas* seedlings subjected to 6 days of salinity (100 mM NaCl) or control treatment. Upper-case letters represent significantly different means between treatments within the same species, and lower-case letters represent significantly different means between species and with the same treatment (Tukey's test; $P < 0.05$).

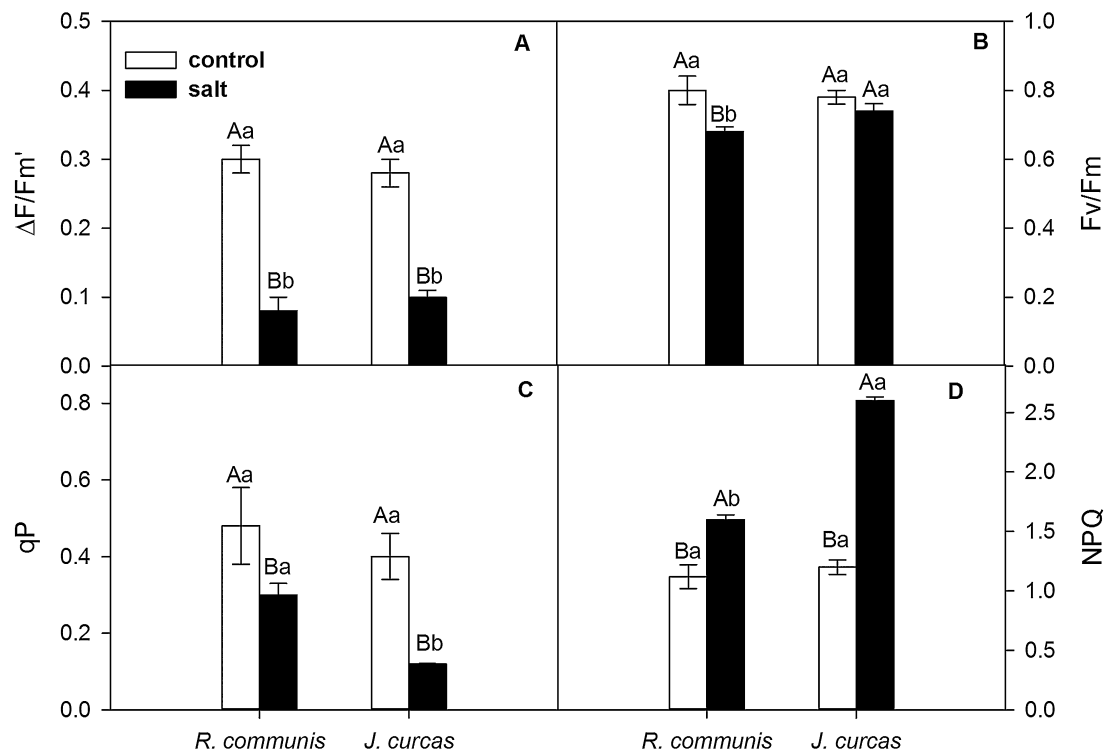


Fig. 2. Effective quantum yield of PSII (A), potential quantum yield of PSII (B), photochemical quenching (C) and CO_2 intercellular partial pressure (D) in *R. communis* and *J. curcas* seedlings subjected to 6 days of salinity (100 mM NaCl) or control treatment. Upper-case letters represent significantly different means between treatments within the same species, and lower-case letters represent significantly different means between species and with the same treatment (Tukey's test; $P < 0.05$).

quenching (NPQ) was slightly increased by salt stress in *R. communis* and strongly in *J. curcas* (Fig. 2D).

The photosynthetic efficiency parameters estimated from the modeling of P_N/C_i and P_N/PPFD curves were significantly affected by

salt treatment in both species (Table 3). The maximum photosynthetic rate ($P_{N\text{max}}$) decreased similarly in both species in response to salinity. The maximum carboxylation rate of Rubisco (V_{cmax}) decreased more prominently in *J. curcas*, whereas the maximum

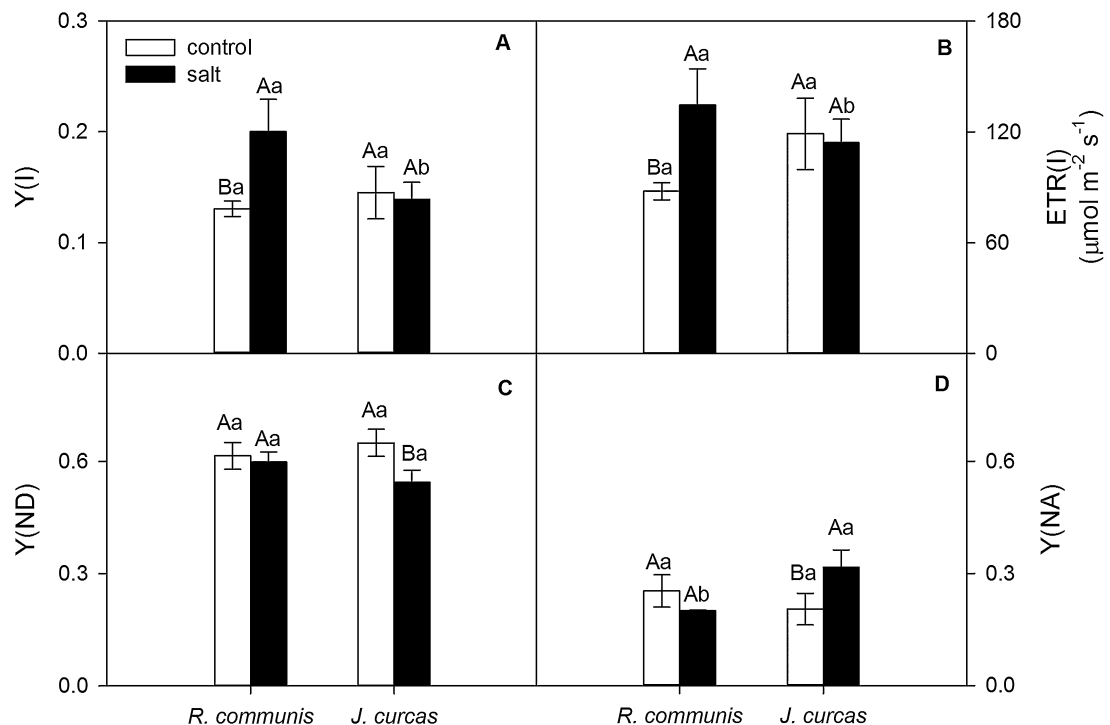


Fig. 3. Effective quantum yield of PSI (A), electron transport rate at PSI (B), non-photochemical donor site limitation of PSI (C) and non-photochemical acceptor site limitation of PSI (D) in *R. communis* and *J. curcas* seedlings subjected to 6 days of salinity (100 mM NaCl) or control treatment. Upper-case letters represent significantly different means between treatments within the same species, and lower-case letters represent significantly different means between species and with the same treatment (Tukey's test; $P < 0.05$).

Table 3

Parameters calculated from the P_N -PPFD and P_N - C_i response curves. P_{Nmax} is the maximum CO_2 assimilation rate; R_d is respiration in the light; R_N is respiration in the dark; V_{cmax} is the maximum carboxylation rate of Rubisco; J_{max} is the maximum photosynthetic electron transport rate; and CEF is the relative cyclic electron flux in *R. communis* and *J. curcas* seedlings subjected to 6 days of 100 mM NaCl (salt) or control treatment. Upper-case letters represent different means among treatments for the same species, and lower-case letters represent different means among species for the same treatment (Tukey's test; $P < 0.05$).

	<i>R. communis</i>		<i>J. curcas</i>	
	Control	Salt	Control	Salt
P_{Nmax} ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	29.04Aa	8.63Ba	21.22Ab	6.01Ba
R_N ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	1.78Ba	2.43Aa	1.63Aa	1.94Ab
V_{cmax} ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	124Aa	86Bb	111Ab	28Ba
J_{max} ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	132Aa	52Ba	92Ab	32Bb
P_R ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	3.53Ba	9.52Ab	4.32Aa	5.21Ba
CEF (r.u.)	0.48Bb	2.32Aa	0.76Ba	1.85Ab

electron transport rate (J_{max}) decreased similarly in both species. Respiration in the dark (R_N) increased in *R. communis* but did not change in *J. curcas*. The estimated photorespiration rate increased greatly in *R. communis* (by two-fold) and slightly in *J. curcas*. The CEF was increased in both species under salt stress but *R. communis* displayed higher values than *J. curcas* under salinity (Table 3).

The effective quantum yield of PSI [Y(I)] and the apparent electron transport at PSI (ETR I) showed the same pattern in both species under salt stress (Fig. 3). Y(I) and ETR I increased in *R. communis* but did not change in *J. curcas* plants (Fig. 3A and B). The non-photochemical donor site limitation of PSI [Y(ND)] was not affected in *R. communis* but was decreased in *J. curcas*. However, the non-photochemical acceptor site limitation of PSI [Y(NA)] was slightly decreased in *R. communis* and, in contrast, was slightly increased in *J. curcas* (Fig. 3C and D).

Salt stress significantly decreased the initial activity of Rubisco in *J. curcas*, but no change was noted in *R. communis* (Fig. 4A). Salinity did not affect the activation state of Rubisco in both studied species (Fig. 4B). GO activity increased by three-fold and by 71% in *R. communis* and *J. curcas*, respectively (Fig. 5B). CAT activity increased in both species under salinity but exhibited values that were almost two-fold higher in *R. communis* than in *J. curcas* (Fig. 5A).

Nitrate and ammonia assimilation and nitrate and ammonium concentration

Salt stress increased the leaf nitrate content in *J. curcas* plants but did not change the leaf nitrate content in *R. communis* (Fig. 6A). The leaf ammonium content was increased in *R. communis* but was decreased in *J. curcas* (Fig. 6B). Leaf NR activity increased by 52% in *R. communis* but was decreased by 67% in *J. curcas* (Fig. 6C). GS activity did not change in both species under salinity, but *R. communis* presented activity that was approximately five-fold higher than that observed for *J. curcas* (Fig. 6D).

Discussion

In the current study, we demonstrated that *R. communis* young plants are less sensitive to salt stress than *J. curcas*. The better salt acclimation exhibited by the former species is clearly demonstrated by the absence of negative effects caused by salt on leaf growth (dry weight) and the maintenance of cellular integrity, as indicated by the leaf electrolyte leakage. These stress parameters are good indicators of salt acclimation/tolerance in glycophytes. Those favorable indicators were associated with physiological mechanisms that supported better growth in *R. communis* under salinity such as higher CO_2 assimilation and nitrate reductase activity in leaves. In addition, *R. communis* displayed lower decreases in Rubisco activity,

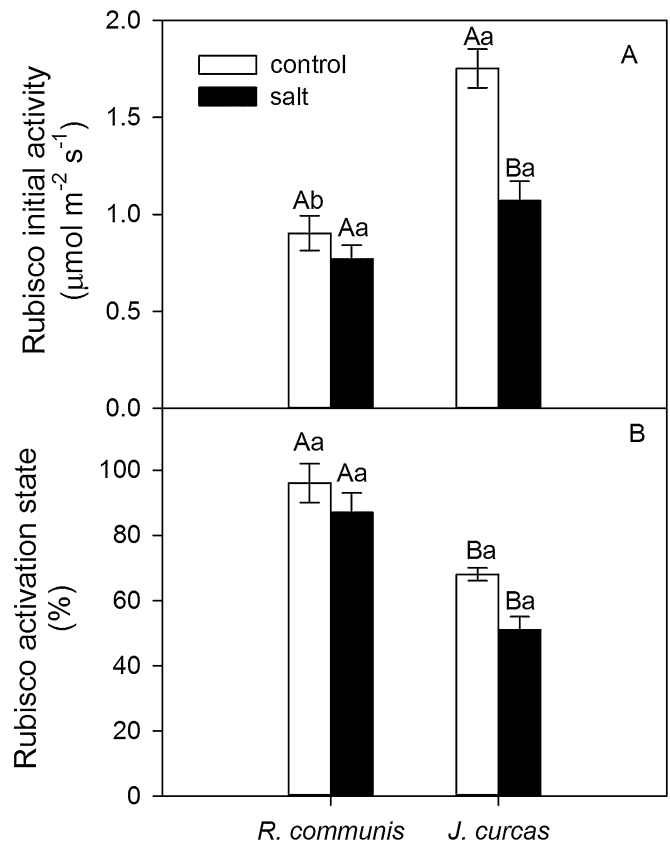


Fig. 4. Changes in initial Rubisco activity (A) and Rubisco activation state (B) in *Ricinus communis* and *Jatropha curcas* seedlings subjected to 6 days of 100 mM NaCl (salt) or control treatment. Upper-case letters represent significantly different means between treatments within the same species, and lower-case letters represent significantly different means between species and with the same treatment (Tukey's test; $P < 0.05$).

demonstrating a lower metabolic limitation of photosynthesis than *J. curcas*. Under salt stress *R. communis* exhibited higher activity of PSII (photochemical quenching), compared with *J. curcas* and increased effective quantum yield and electron transport rate in PSI, compared with control. In addition, the estimated CEF was higher in *R. communis* plants exposed to salt compared with *J. curcas*. These parameters associated with the CEF, which is an important mechanism for excess energy dissipation under conditions of impairment of CO_2 assimilation (Johnson, 2011) caused by salinity.

Interestingly, the salt-sensitive species *J. curcas*, also exhibited some physiological characteristics favorable to salt acclimation compared with *R. communis*. This species presented higher K/Na ratios and higher K^+ concentration in leaves under salinity. In most glycophytes, this first parameter is very important for salt tolerance (Munns and Tester, 2008). However, in the current study, the Na^+ content in leaves did not reach high and toxic levels in both species (Silva et al., 2011). It was previously demonstrated that *J. curcas* accumulates high concentrations of Na^+ in leaves in presence of 100 mM NaCl after only two weeks (Silva et al., 2011). These results are in agreement with the obtained data in this current study, evidencing that after six days of salt exposure both *J. curcas* and *R. communis* seedlings did not reach toxic Na^+ levels in their leaves.

Previous studies with *J. curcas* young plants have suggested that this species is salt sensitive and suffers irreversible damage on the photosynthetic apparatus after a two-week exposure to 100 mM NaCl. *J. curcas* can exhibit a leaf Na^+ -exclusion mechanism only during approximately one week when the osmotic phase is predominant (Silva et al., 2011). Conversely, some studies have

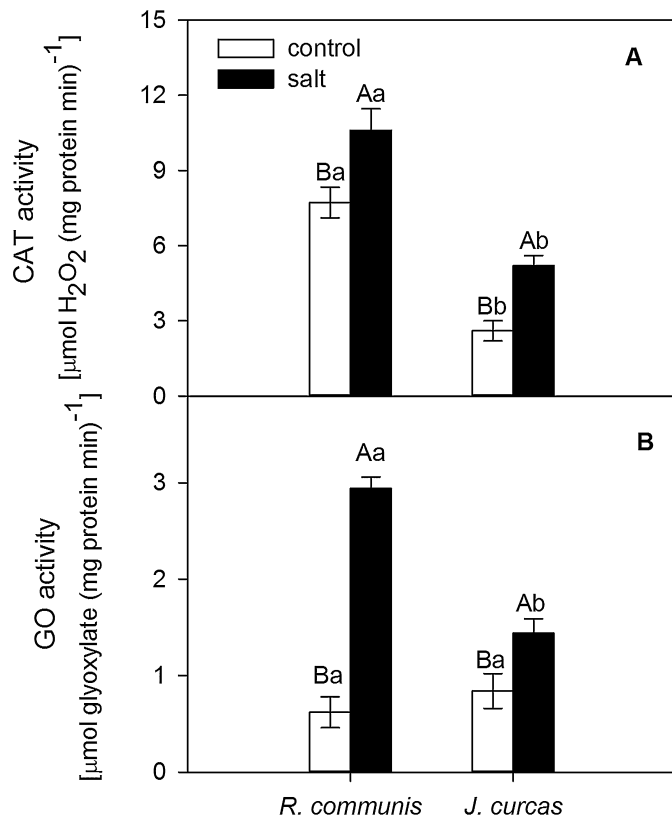


Fig. 5. Catalase activity (A) and glycolate oxidase activity (B) in *Ricinus communis* and *Jatropha curcas* seedlings subjected to 6 days of 100 mM NaCl (salt) or control treatment. Upper-case letters represent significantly different means between treatments within the same species, and lower-case letters represent significantly different means between species and with the same treatment (Tukey's test; $P < 0.05$).

demonstrated that *R. communis* is a salt-tolerant species (Pinheiro et al., 2008; Li et al., 2010; Sun et al., 2013). Because both species in this present study exhibited similar leaf- Na^+ concentrations, and *R. communis* presented lower K/Na ratios, one central question can be raised here: which others physiological characteristic(s) would have contributed to salt tolerance in *R. communis*? Under salinity, this species exhibited a great stimulation in photorespiration. This situation is particularly important under high salinity when CO_2 assimilation is impaired and electron excess is accumulated in thylakoid. In this study the higher photorespiratory rate in *R. communis* was clearly demonstrated by the gas exchange parameter and biochemical indicators (activities of GO, GS and CAT).

The nitrite and ammonia assimilation in photosynthesizing leaves contribute as a strong alternative sink for the (PET) chain, converting the reduced ferredoxin in its oxidized form that can accept new electrons from PSII and PSI. In the present study, the increase in the NR activity in *R. communis* leaves under salinity, associated with maintenance of high GS activity, strongly suggest that nitrate and ammonia assimilation might have contributed to mitigate the negative effects caused by salinity on the photosynthesis.

The salt tolerant species *R. communis* exhibited intriguing results and to some extent, conflicting with literature in terms of nitrate reductase activity and photosynthetic CO_2 assimilation. In general, these two processes act coordinately in most species under normal conditions. NaCl stimulated nitrate reductase activity but the photosynthesis was strongly impaired in *R. communis* leaves. Some plant species like soybean has a constitutive form of NR that is more independent on nitrate transport from roots to leaves affected by transpiration than NO_3^- -inducible forms (Silveira et al., 2001). Possibly, in salt-treated *R. communis* the levels of NR protein, NO_3^- and cytosolic NADH were favorable to stimulate nitrate reductase activity but not for Rubisco expression and activity. This question is interesting and further studies are needed for elucidation. In addition to stomatal limitation, *J. curcas* also presented biochemical limitations in photosynthesis, as shown by

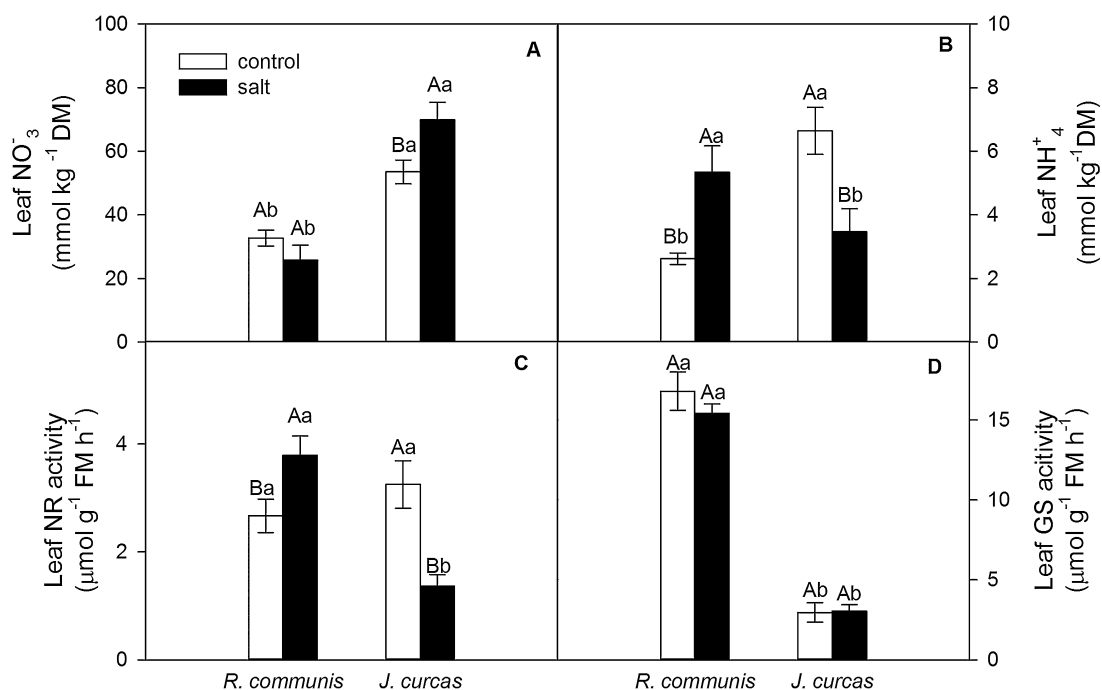


Fig. 6. Changes in leaf (A) NO_3^- , (B) NH_4^+ content, (C) nitrate reductase and (D) glutamine synthase activities in *Ricinus communis* and *Jatropha curcas* seedlings subjected to 6 days of 100 mM NaCl (salt) or control treatment. Upper-case letters represent significantly different means between treatments within the same species, and lower-case letters represent significantly different means between species and with the same treatment (Tukey's test; $P < 0.05$).

the increases in C_i values, decrease in Rubisco activity *in vitro* and greater reduction in V_{cmax} and J_{max} compared with *R. communis* plants exposed to salt stress. Conversely, *J. curcas* plants presented an efficient energy excess dissipation mechanism as heat by NPQ under salt stress possibly protecting the PSII against photodamage and photoinhibition. The increase in NPQ is an important mechanism for photoprotection in plants, dissipating the excess energy as heat, avoiding damage at the PSII core (Takahashi and Badger, 2011).

In this study, we reported a non-classical mechanism involved with salt tolerance based on three processes related with excess energy dissipation from the photosynthetic electron transport chain. The salt tolerance of *R. communis* compared with *J. curcas* is associated with higher photorespiration, nitrate assimilation and cyclic electron flow. These processes might contribute to protection against photodamage, photoinhibition and photo-oxidative stress under salinity.

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